

# Seasonal influence of physico - chemical parameters on Algal diversity of an urban lake in Hyderabad, India

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#### Abstract

Limnological studies of an urban lake, Nadimi Cheruvu, Hyderabad were carried out monthly over a period of two years. The present paper highlights seasonal variation between various physico-chemical parameters and their influence on algal diversity and distribution in this water body. During the study period, the algal flora fluctuated between 31,000 nos/L (in winter) - 70,300 nos/L (in summer) and phytoplankton diversity of the lake revealed the presence of 88 algal species that belongs to 41 genera. Chlorophyceae is the most dominant group while Euglenophyceae is least represented group. Chlorophyceans exhibited high multiplication in summers and low in winters. Cyanophyceaen minima occurred in monsoon and its maxima in summers. Dinophyceaen were recorded more in number during winter months and less during summer months. Bacillariophyceaens found to be maximum in summer and minimum in monsoon season in the first year and during winter season in the second year. Pearson's correlation analysis has been calculated to find out the relative importance between physico-chemical variables and phycoflora. Chlorophyceae and Bacillariophyceae exhibited a marked degree positive correlation with water temperature, pH, TH and calcium; Dinophyceae expressed a marked degree negative correlation with water temperature; Cyanophyceae displayed a moderate degree positive correlation with DO, COD, BOD, organic matter, phosphates, nitrates, silicates, magnesium, bicarbonates, carbonates, alkalinity and EC.

Keywords: Nadimi Cheruvu, Safilguda Lake, Limnological studies, Seasonal variation

### 1. Introduction

Algae, one amongst the diverse groups of living organisms are distributed in different types of water habitats. The phytoplankton community plays a key role in aquatic ecosystems and provides basic requirements such as carbon fixation, oxygen production and as food generation (Fathi et al. 2001; Khan 2003). Phytoplankton communities exhibit rapid succession shifts in response to environmental changes. The complex interaction between physical, chemical and biological factors in aquatic environment influences the diversity, dynamics and distribution of phytoplankton population (Smayda, 1980).

Phytoplankton community structure, composition and species diversity in aquatic ecosystem are determined by several limnological factors (Sin et al., 1999; Ghosh et. al., 2012). Phytoplankton follows a fairly recognizable annual cycle of growth, but sometimes the synchrony in their normal



annual cycle is disrupted by the explosive growth of some species (Vaulot, 2001). The occurrence and abundance of diverse algae vary seasonally and their study provides a relevant focus for research on eutrophication of water bodies and its adverse impact on aquatic life (Pranita and Papiya, 2018).

In India, many studies on the influence of physico-chemical factors on algal diversity of Freshwater bodies were by various scholars (Pruthi, carried 1933; Ganapati, 1960; Zafar, 1967; Mary Cynthia 1980; Venkateswarlu and Manikya Reddy, 1985; Hosmani et, al, 1999; Annie sheron, 2003; Kumawat and Jawale, 2004; Banakar et. al., 2005; Veereshakumar and Hosmani, 2006; Ruchira and Rachana, 2009)

The present investigation focuses on a comprehensive study of seasonal influence of various limnological variables on phytoplankton diversity, their species composition, population density and their distribution in an urban lake of Hyderabad. Further, The Pearson correlationcoefficient was used to establish the relationship among various environmental variables with phytoplankton density. The field work was spread over a period of two years from March 2004 to February 2006.

#### 2.0 Materials and Methods

2.1 Study area: Founded in 1591 on the banks of the river Musi, the city of Hyderabad is located in the heart of the Deccan plateau at 1700 feet above sea level. The city is bestowed with enthralling topography and beautiful landscapes decked with amazing prehistoric formations and natural lakes. According to Census of India, 2011, it is the 4th populous city in India and a multitude of factors such as population growth, industrialization, construction activities put this city under continuous stress. Once known as 'City of Lakes' lost most of its water bodies to pollution and other human centric developmental activities.



The urban waterbody chosen for the present study is 'Nadimi Cheruvu' which is also known as 'Safilguda Lake'

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situated in Malkajgiri municipality of Hyderabad Metropolitan area and lies in the coordinates of 17.2730°N, 78.3220°E.



According to the state gazette the water spread area of the lake 50 years before was FTL of 537 Mts and spread over 3.95 square miles and was a good water source of irrigation and drinking. Due to the unprecedented population growth and industrialization in the last few decades the extent of water spread is reduced to 27.5 Acres (Annie Sheron, 2003). In 2001 this the lake was restored, Demarcated, dredged and fenced under 'Lake Conservation Programme' initiated by Hyderabad Urban Development Authority (HUDA). А sewerage treatment plant (STP) of capacity 0.6 MLD, has been constructed to treat the dry weather flow of sewage and it is functional since January' 2003.

**2.3: Collection and analysis of samples :** Water samples were collected for physico chemical and phycological analysis from these water bodies at monthly intervals for two years i.e. from February 2004 to January 2006. .

Samples were collected in plastic Jerry cans of two liters capacity at four different spots of the lake. All the samples were kept in the icebox and were for brought to the laboratory enumeration of quantitative and qualitative analysis. Parameters like temperature, Electrical Water pН, Conductivity (EC) and Dissolved Oxygen (DO) were recorded at the time of collection in the field itself by using YSI Environmental Field probes. Variables like Total solids (TS), Total Dissolved Solids (TDS), Total suspended Solids (TSS), alkalinity, Silicates, Nitrates, Phosphates, Chemical Oxygen Demand (COD) and Biological Oxygen Demand analysed (BOD) have been using methods described by APHA (1998); Carbonates, Bicarbonates, Total Hardness, Organic Matter, Calcium, Magnesium, Chlorides were analysed by standard methods described by Wilcox and Hatcher (1950).

Qualitative and quantitative estimation of Phytoplanktonic algae was taken into consideration under phycological parameters. For algal studies, Surface water samples were collected in one liter sterilized clean high density polyethylene kept bottles and were in the sedimentation columns after adding 2-3 ml of Lugol's solution. The samples were set aside undisturbed for about 4 weeks for a complete setting of the organisms and finally the sample is concentrated to Detailed 100 ml. analyses of phytoplankton populations are done by estimating the numbers in each species. The algal samples were identified under the binocular compound microscope and counting was done by Drop count method (Trivedi and Goel, 1984). Motile algae was identified from fresh samples. Qualitative identification of phytoplankton forms up to species level was done with the help of monographs 1959; Suxena (Desikachary, and Venkateswarlu, 1966; Phillipose, 1967; Anand, 1998 and Rai, 2006).

## 3.0 Results and Discussion:

### 3.1 Physico chemical parameters

Results of the physico- chemical analysis of lake water collected are represented in Tables 1 & 2. For the convenience of discussion, three seasons viz. Summer, Monsoon and Winter are considered to compute the water quality trends in two years of study. Monthly Variations in Physico-Chemical Parameters at Nadimi Cheruvu were represented in Fig. 1.

Throughout the study period the colour of the Safilguda Lake water was found to



be light greenish in colour but during the summers it was found to be green in colour emitting algal odour. The water temperature ranged between 21.6°C (Jan, 2006) - 32.6°C (May 2004) with a mean temperature of 26.5°C. Seasonally, water temperature has been recorded high during the summer (28.9°C) and low during winter (23.3°C). Increase in water during temperature summer and decrease during winter seasons were reported by many authors (Sakhre and Joshi, 2003 and Garg et al., 2009). pH of the lake water was always alkaline in nature, ranging from a maximum of 10.9 against a minimum of 7.27 with an average of 8.7. Season wise, pH was more in summers followed by monsoon and less in winter.

EC was recorded maximum in summer (1540.8 mhos/cm) and minimum in winter (1286 mhos/cm). Low average turbidity values were recorded during monsoon (45.8 NTU) and high in summer (57.4 NTU). Summer maxima of EC and turbidity could be due to decomposition organic matter, of decrease in water level due to evaporation. The results are in agreement with the findings of Sulabha and Prakasham (2006). Alkalinity ranged between 72 mg/L - 360 mg/L and season wise, the average total alkalinity was found to be high during summer (221 mg/L) and low in monsoon (135 mg/L) which can be attributed to low water level and high evaporation in lake water. Similar results were reported by Sakhre and Joshi (2003)

Total hardness was recorded low in monsoon (209 mg/L) and high in summer (300 mg/L). Carbonates were observed in lower concentrations when compared to bicarbonates. TSS and TDS was optimum in rainy season and minimum in winter. This indicates that the dissolved were materials of allochthonous in origin which were brought into the lake with surface run off and leaching of bund soil. Similar opinion has been expressed by Hosmani et, al, (1999) and Narain and Chauhan (2002). High amounts of calcium recorded in summer (231mg/L) and low amounts in monsoon (149.6mg/L). Magnesium content was more in summer (68.9 mg/L) and less in monsoon (59.1mg/L). During the entire period of study Magnesium was observed always lower than calcium concentration and calcium richness of the water could be attributed to the high concentrations of bicarbonates and it is in conformity with the investigations of Janardhan Rao (1982) and Unni et al., (1992).

Low Chlorides and Silicates concentration is recorded in monsoon (83 & 40 mg/L) and Chlorides concentration is more in summer (123mg/L) where as Silicates in winter (56.8 mg/L). Nitrates and Phosphates were found in low concentration in the range of 0.41mg/L -2.8 mg/L and 0.65 mg/L – 4.58 mg/L. It may be due reduced load of these nutrients as the water is treated by the sewerage treatment plant before letting into the lake.

Lake water was well aerated. High concentration of DO was noticed throughout the period of investigation and it never reached less than 9.1 mg/L and it was always higher than organic matter. The free CO2 was found to be high during winter (13 mg/L) and low during summer (8.1 mg/L). BOD & COD found to be high in summer (32.9 mg/L & 85 mg/L) and low during winter (24.3 mg/L & 76.5 mg/L). Singh et. al.,(1999) and Banakar (2006) who concluded that high temperatures enhances the

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biological activities resulting in the increase of BOD.

Concentration of EC, TS, TDS, alkalinity, carbonates, bicarbonates, Mg, Chlorides, Silicates, Nitrates, Phosphates, BOD and COD has increased from 1<sup>st</sup> year to 2<sup>nd</sup> year of study period.

3.2 Phycoflora During the study period, the algal flora fluctuated between 31,000 nos/L (in winter) - 70,300 nos/L (in summer) and phytoplankton diversity of the lake revealed the presence of 88 algal species that belongs to 41 genera and the order of dominance of various algal Chlorophyceae groups was > Cyanophyceae> Bacillariophyceae> Euglenophyceae Dinophyceae> that occupied a portion of 57%, 25.2%, 12.7%, 4.7%, and 0.4% respectively in the total bulk. phytoplankton Algal species recorded during the study period were listed in table 3 and Monthly variation in algal flora is represented in Fig. 2.

Chlorophyceae was observed as a stable and dominant algal community (38 species belonging to 20 genera) and seasonally, it exhibited high multiplication in summers and low in winters. It is represented by the members belong to order Volvocales, Chlorococcales and Desmids. Volvocales multiplied profusely during monsoon and meagrely during winter and is predominantly represented, Chlorogonium, Gloeocystis and Pandorina. Chlorococcales multiplied copiously in summers and formed a peak

in May of both the years of study period. Similar findings was made by Kumawat and Jawale (2004). Desmids started multiplying during January and February months and reached a peak towards April and their populace low during November and December months. *Staurastrum* species were found to be more in number during monsoon.

**Cyanophyceae is** represented by 9 genera and 18 species and exhibited a wellmarked periodicity by showing a winter peak that extended to early summer with a distinct monsoon peak. The summers were dominated by *Oscillatoria curviceps, Merismopedia glauca*.

Bacillariophyceans were fairly represented (11 genera and 27 species) and their number is maximum in summer and minimum in monsoon season of the first year and winter season of the second year.

Euglenophyceae is the least represented group (4 species belonging to 2 genera). Euglenophyceaens number is more in monsoon and they are not recorded in the 1<sup>st</sup> year winter season and their presence in the 2<sup>nd</sup> year winter is very low.

**Dinophyceae is** represented by 2 genera and 4 species and seasonally, they were recorded more in number during winter months and less during summer months. *Glenodinium* spp. multiplied and found a peak in monsoon and *Ceratium* sp. was observed in winters in few numbers when water temperatures dropped to 20°C - 21°C

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#### 3.3 Influence of physicochemical parameters on phytoplankton distribution

temperature plays The water an important role in controlling the occurrence and abundance of phytoplankton. Summers showed maximum algal density and Minimum density of phytoflora was recorded during monsoon and it was due to low pH, low DO, turbid water similar observations were made by Kumawat and Jawale (2004) and Banakar, (2006).

High temperature, bright sunshine, slight alkaline pH, low calcium and low organic matter concentration appears to favour the abundance of desmids. Zafar (1967) indicated that calcium enrichment of water results in the general paucity of desmid population. Similar trend was observed in the present study where less calcium concentration (182mg/L) supported good desmid population.

Warm temperatures, total hardness supported luxuriant growth of Cyanophyceans. This is in conformity



with the findings of Banakar et al. 2005 and Veereshakumar and Hosmani, 2006.

It is difficult to assess the periodicity of Euglenoids because some species appear abruptly, sometimes almost instantly and sometimes remain unseen from the surface water for long duration.

Silicate content fluctuated verv effectively with the growth pulse of diatoms. The peak values in silicates were followed by diatom peak. This clearly indicates that pH above 9.0 is unfavourable for silica to dissolve in lake water. This is in conformity with the findings of Ruchira and Rachana (2009). Phosphate rich waters and high DO concentration supported luxuriant growth of diatoms. Venkateswarlu and Manikya Reddy (1985) and Banakar et. al. (2005) have also made similar findings.

Dinophyceans preferred low temperature, low organic matter, nonturbid conditions with low electrical conductivity and high DO concentrations. This is in conformity with the findings of Mary Cynthia (1980).

### 3.4 Correlation Studies

Pearson's correlation analysis has been calculated to find out the relative importance between various physicochemical variables, between phycoflora and between physico-chemical variables and phycoflora

3. 4a Correlation between physicochemical parameters: The water temperature revealed marked degree positive correlation with pH and a marked degree negative correlation with DO; pH exhibited a marked degree positive correlation with carbonates and total hardness; EC showed a marked degree positive correlation with COD, BOD and free CO<sub>2</sub>; TS and TDS expressed a high positive correlation with each other; Carbonates expressed a marked degree positive correlation with EC, pH and BOD; Total hardness and calcium displayed a high degree of positive correlation; Nitrates expressed marked degree positive correlation with COD; DO expressed a high negative correlation with water temperature; COD and BOD showed a high degree positive correlation with each other, a marked degree positive correlation with EC.

3. 4b Correlation between physicochemical parameters and phycoflora: Chlorophyceae and Bacillariophyceae exhibited a marked degree positive correlation with water temperature, pH, TH calcium; and With DO Chlorophyceae showed a marked degree Dinophyceae negative correlation; expressed a marked degree negative correlation with water temperature; Cyanophyceae displayed a moderate degree positive correlation with DO, COD, BOD, organic matter, phosphates, nitrates. silicates, magnesium, bicarbonates, carbonates, alkalinity and EC.

3.4c Correlation between phycoflora: Chlorophyceae and Bacillariophyceae exhibited high positive correlation where as Chlorophyceae and Cyanophyceae exhibited a low degree negative correlation. With Chlorophyceae and Euglenophyceae, Dinophyceae displayed a moderate degree negative correlation and with Bacillariophyceae it expressed a low degree negative correlation.

The present study affirms that plaktonic diversity and distribution can be best explained when environmental



factors jointly influence. It can be concluded that the seasonal changes in the physico chemical factors either directly or indirectly influence the algal diversity.

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| Table 1: Physico-chemical parameters of Nadimi Chernyn (March '04 - February '05) |                      |        |            |        |        |        |            |        |        |            |        |        |        |        |
|---|----------------------|--------|------------|--------|--------|--------|------------|--------|--------|------------|--------|--------|--------|--------|
| S.No  | PARAMETERS           | Mar-04 | Apr-<br>04 | May-04 | Jun-04 | Jul-04 | Aug-<br>04 | Sep-04 | Oct-04 | Nov-<br>04 | Dec-04 | Jan-05 | Feb-05 | Avg    |
| 1.  | Water Temp. (°C)     | 25.4   | 28.6       | 32.6   | 28.6   | 28.1   | 28         | 28.2   | 26.1   | 25.2       | 22.5   | 24     | 24.4   | 26.8   |
| 2.  | pH                   | 8.4    | 8.9        | 10.1   | 9.8    | 8.2    | 8.61       | 8.8    | 8.2    | 7.38       | 7.9    | 7.27   | 8.4    | 8.5    |
| 3.  | Turbidity (NTU)      | 47.8   | 42         | 52     | 65.1   | 73     | 39.2       | 33     | 25     | 60         | 57.5   | 42     | 52.7   | 49.1   |
| 4.  | EC (mhos/cm)         | 1204   | 1213       | 1609   | 1348   | 1224   | 1035       | 932    | 1067   | 920        | 1002   | 908    | 1012   | 1122.8 |
| 5.  | TS                   | 734    | 647        | 776    | 953    | 900    | 990        | 849    | 915    | 755        | 789    | 850    | 903    | 838.4  |
| 6.  | TDS                  | 633    | 616        | 722    | 876    | 817    | 925        | 784    | 827    | 742        | 756    | 765    | 846    | 775.8  |
| 7.  | TSS                  | 101    | 31         | 54     | 77     | 83     | 100        | 65     | 88     | 13         | 33     | 85     | 57     | 65.6   |
| 8.  | Alkalinity           | 266    | 260        | 183    | 75     | 97     | 214        | 113.5  | 183    | 169        | 149    | 72     | 100    | 156.8  |
| 9.  | Carbonates           | 16     | 22         | 24     | 12     | 0      | 0          | 22     | 26     | 0          | 0      | 0      | 16.2   | 11.5   |
| 10.   | Bicarbonates         | 178    | 143        | 125    | 63     | 70     | 141        | 70     | 114    | 70         | 64     | 170    | 80     | 107.3  |
| 11.   | Total Hardness       | 240    | 252        | 352    | 344    | 200    | 180        | 180    | 240    | 266        | 244    | 210    | 216    | 243.7  |
| 12.   | Calcium              | 190    | 208        | 288    | 256    | 156    | 129        | 120    | 177    | 200        | 200    | 136    | 174    | 186.2  |
| 13.   | Magnesium            | 50     | 44         | 64     | 88     | 44     | 51         | 60     | 63     | 66         | 44     | 74     | 42     | 57.5   |
| 14.   | Chlorides            | 161    | 112        | 152    | 127    | 75     | 67         | 63     | 72     | 108        | 88     | 100    | 95     | 101.7  |
| 15.   | Silicates            | 40     | 60         | 72     | 36     | 24     | 20         | 32     | 28     | 40         | 52     | 62     | 56     | 43.5   |
| 16.   | Nitrates             | 0.41   | 1.1        | 1.6    | 0.54   | 0.41   | 0.42       | 1.08   | 0.61   | 1.58       | 0.72   | 1.3    | 2.2    | 1.0    |
| 17.   | Phosphates           | 0.65   | 0.9        | 0.83   | 1.1    | 0.65   | 0.98       | 0.65   | 0.68   | 1.08       | 1.11   | 3.48   | 4.58   | 1.4    |
| 18.   | DO                   | 8.1    | 7.8        | 5.2    | 6.8    | 8.8    | 8.1        | 8.4    | 8.1    | 9.2        | 11.8   | 10.6   | 10.8   | 8.6    |
| 19.   | Free CO <sub>2</sub> | 10.6   | 3.8        | 6.2    | 8      | 12     | 6          | 6      | 10     | 10         | 12.8   | 14     | 12     | 9.3    |
| 20.   | Organic Matter       | 2.25   | 3          | 4.8    | 2.4    | 1.2    | 2.1        | 2      | 1.7    | 2          | 2.8    | 1.5    | 2.8    | 2.4    |
| 21.   | COD                  | 34     | 46         | 54     | 60     | 38     | 50         | 48     | 55     | 62         | 42     | 80     | 80     | 54.1   |
| 22.   | BOD                  | 18     | 16         | 22     | 30     | 14     | 18         | 20     | 19     | 20         | 12     | 32     | 28     | 20.8   |

All the parameters from 5 – 22 are in mg/L  $\,$ 

Table 2: Physico-chemical parameters of Nadimi Cheruvu (March '05 - February '06)

| S.No | PARAMETERS           | Mar-05 | Apr-<br>05 | May-05 | Jun-05 | Jul-05 | Aug-<br>05 | Sep-05 | Oct-05 | Nov-<br>05 | Dec-05 | Jan-06 | Feb-06 | Avg    |
|------|----------------------|--------|------------|--------|--------|--------|------------|--------|--------|------------|--------|--------|--------|--------|
| 1.   | Water Temp. (°C)     | 27.3   | 28.4       | 31.4   | 28.7   | 27.7   | 27.2       | 28     | 25.4   | 24.5       | 22.3   | 21.6   | 22     | 26.2   |
| 2.   | pH                   | 10.1   | 10.4       | 10.9   | 9.1    | 8.5    | 8.62       | 8.46   | 8.2    | 8.12       | 7.9    | 7.57   | 8.4    | 8.9    |
| 3.   | Turbidity (NTU)      | 67.3   | 80.6       | 48     | 56     | 44.2   | 48.3       | 46     | 58     | 67         | 41.5   | 49.3   | 58.9   | 55.4   |
| 4.   | EC (mhos/cm)         | 1734   | 1762       | 1789   | 1667   | 1780   | 1620       | 1724   | 1688   | 1431       | 1560   | 1440   | 2013   | 1684.0 |
| 5.   | TS                   | 973    | 850        | 981    | 1100   | 1225   | 1123       | 1032   | 978    | 806        | 814    | 872    | 897    | 970.9  |
| 6.   | TDS                  | 899    | 785        | 903    | 1000   | 1126   | 1008       | 925    | 903    | 713        | 727    | 785    | 800    | 881.2  |
| 7.   | TSS                  | 74     | 65         | 78     | 100    | 109    | 115        | 107    | 85     | 93         | 87     | 87     | 97     | 91.4   |
| 8.   | Alkalinity           | 250    | 290        | 360    | 87     | 90     | 81         | 115    | 183    | 189        | 149    | 183    | 240    | 184.8  |
| 9.   | Carbonates           | 40     | 54         | 28     | 32     | 18     | 14         | 28     | 22     | 28         | 0      | 12.4   | 37.8   | 26.2   |
| 10.  | Bicarbonates         | 250    | 179.6      | 131    | 64.3   | 64     | 73         | 86.3   | 91.7   | 92         | 108    | 160    | 182    | 123.5  |
| 11.  | Total Hardness       | 228    | 280        | 380    | 320    | 200    | 200        | 270    | 200    | 200        | 180    | 270    | 230    | 246.5  |
| 12.  | Calcium              | 156    | 218        | 280    | 249    | 120    | 138        | 228    | 129    | 148        | 132    | 212    | 132    | 178.5  |
| 13.  | Magnesium            | 72     | 62         | 100    | 71     | 80     | 62         | 42     | 71     | 52         | 48     | 58     | 98     | 68.0   |
| 14.  | Chlorides            | 98     | 100        | 103    | 134    | 120    | 70         | 98     | 100    | 107        | 125    | 130    | 142    | 110.6  |
| 15.  | Silicates            | 30     | 68         | 72     | 52     | 52     | 58         | 46     | 60     | 62         | 68     | 68     | 46     | 56.8   |
| 16.  | Nitrates             | 0.9    | 3.5        | 2.1    | 2.2    | 2.2    | 0.44       | 2.8    | 2.1    | 2          | 1.56   | 1.28   | 2      | 1.9    |
| 17.  | Phosphates           | 2.86   | 2.55       | 2.48   | 3.3    | 2.17   | 1.87       | 1.91   | 1.55   | 1.86       | 1.72   | 2.03   | 2.24   | 2.2    |
| 18.  | DO                   | 9.6    | 9.6        | 8.2    | 7.1    | 8.7    | 8.6        | 8.4    | 8.8    | 10.1       | 11     | 14     | 11     | 9.6    |
| 19.  | Free CO <sub>2</sub> | 4      | 8          | 10     | 14     | 10     | 6          | 12     | 6.8    | 8.4        | 18     | 13.2   | 15.2   | 10.5   |
| 20.  | Organic Matter       | 3.2    | 2.8        | 3.8    | 2.8    | 2.8    | 1.9        | 2      | 2.6    | 3.8        | 3      | 3.4    | 3.5    | 3.0    |
| 21.  | COD                  | 116    | 158        | 120    | 92     | 112    | 122        | 100    | 130    | 88         | 100    | 60     | 100    | 108.2  |
| 22.  | BOD                  | 42     | 62         | 48     | 25     | 30     | 38         | 30     | 30     | 22         | 30     | 22     | 32     | 34.3   |

All the parameters from 5 – 26 are in mg/L

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#### Table 3: List of Algal species recorded in Nadimi Cheruvu during study period

Cyanophyceae Aphanocapsa grevillei 1 2 Coelosphaerium kuetzingianum 3 Merismopedia glauca

- 4 Merismope dia punctata
- 5 Merismope dia tenuissi ma
- 6 Gloeocapsa coracina
- Microcystis aeruginosa 7
- 8 Microcystis flos- aquae
- 9 Chrococcus elegans
- 10 Chrococcus minimus
- 11 Chroococcus minutus
- 12 Chroococcus turgidus
- 13 Lyngbya spiralis
- 14 Oscillatoria curviceps
- 15 Oscillatoria formosa
- 16 Oscillatoria limmetica
- 17 Oscillatoria princeps
- 18 Nostoc verrucosum Total - 18

### Chlorophyceae

- 19 Pandorina morum
- 20 Chlorogonium elongatum
- 21 Gloeocystis gigus
- 22 Ulothrix subtilissima
- 23 Chlorococcum humicola
- 24 Characium falcatum
- 25 Pediastrum boryanum
- 26 Pediastrum duplex
- 27 Pediastrum ovatum
- 28 Pediastrum simplex
- 29 Pediastrum tetras
- 30 Ankistrodesmus falcatus

- 31 Dictyosphaerium ehrenbergianum 60 32 Kirchneriella hunaris 33 Schroederia setigera
- 34 Tetrae dron c audatum
- 35 Tetrae dron hastatam
- 36 Tetradron incus
- 37 Tetrae dron limneticum
- 38 Tetrae dron minimum
- 39 Tetrae dron muticum
- 40 Tetraedron trigonum
- 41 Tetrae dron tumi dulum
- 42 Crucigenia irregularis
- 43 Crucigenia rectangularis
- 44 Scenedesmus acuminatus
- 45 Scenedesmus bijuga
- 46 Scenedesmus dimorphus
- 47 Cosmarium contractum
- 48 Cosmarium lundelli
- 49 Cosmarium tumidum
- 50 Staurastrum javanicum
- 51 Staurastrum paradoxum Staurastrum tetracerum
- 52 Staurastrum tetrace 53 Euastrum didelta
  - Total 35

#### Euglenophyceae

- 54 Lepocincilus ovum
- 55 Lepocinclis sphagnicola
- 56 Trachelomonas armata
- 57 Trachelomonas volvocina
  - Total 04

#### Bacillariophyceae

- 58 Fragilaria vacucheriae
- 59 Cocconeis placentula

- Achnanthes exigua
- 61 Achnanthes inflata
- 62 Achnanthes microcephala
- 63 Achnanthes minutissima
- 64 Navicula cryptocephala
- 65 Navicula cuspidata
- 66 Navicula pupula
- 67 Navicula hustedtii
- 68 Navicula rhynchocephala
- 69 Gyrosigma kutzingii
- 70 Gomphonema accuminatum
- 71 Gomphonema gracile
- 72 Amphora ovalis
- Amphora veneta 73
- 74 Cymbella affinis
- 75 Cymbella aspera
- 76 Cymbella cuspidata
- 77
- Cymbella cymbiformis
- 78 Cymbella tumida
- 79 Cymbella turgida
- 80 Rhopalodia gibba
- 81 Nitschizia acicularis
- 82 Nitschizia acuminata
- 83 Nitschizia obtusa
- 84 Hantzschia amphioxys Total - 27

#### Dinophyceae

- 85 Glenodinium gymnodinium
- 86 Glenodinium palustre
- 87 Glenodinium cinctum
- 88 Ceratium hirundinella
  - Total-04

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Fig.1: Monthly Variations in Physico-Chemical Parameters at Nadimi Cheruvu